WEEK #3 NOTE: You may choose to complete this assignment on MobyMax using a cell phone
DAY #1 - DIRECTIONS: Read each passage and complete the activities after each.
Pick up an object close to you. How can you describe that object? An example would be a pen. The
physical properties of a pen include its colors and texture: black, metallic, and smooth. All matter in the
universe can be described in some way. When we describe an object, we tell about its physical properties.
Physical properties are things about materials that can be measured or detected by the five senses. Color
texture, and scent are some important physical properties of matter. In physical science, more specific
measurements are also used to describe physical properties. These measurements include mass, weight,
volume, density, buoyancy, conductivity, and solubility. We will learn more about these measurements as
this lesson continues.
Look at the object you just picked up. Use your five senses to list some of its physical
properties below. Be sure to identify your object by name in your response.
Object Name: List 5 senses:
1
2
3.
4.
5.
When listing the physical properties of an object, which of the following might you
include?
□ the taste of the object
□ the texture of the object
the shape of the object
 any noises the object makes
how the object smells
□ the color of the object
Name TWO items that are both spiky and green:
1
2.
Have you ever had your weight measured on a scale at the
doctor's office? Weight is the measure of the pull of gravity on
an object. This means that the weight of something changes
depending on how much gravity is pulling on it. The
pull of gravity is different in different places. It is
stronger at the bottom of the sea than at the top of
a mountain. Therefore, the same object will weigh
more at the bottom of the sea and less at the top of a mountain. The mass of an object is related to its
weight, but mass and weight are not the same. Mass (M) is the measure of the amount of
matter in an object. Mass is measured in grams (g). Mass is measured on a balance by
comparing the object against other objects with known masses. These objects are called
balance weights. An object's weight is a measure of the pull that gravity has on the object.
An object's mass is a measure of the amount of matter in the object. The mass of an object
does not change when gravity changes. However, if you measure two objects in the same place on Earth,
the one with the higher weight will also have a higher mass.
The of an object depends on the pull of gravity.
The of an object depends on how much matter is in the object
If measured in the same place on Earth, mass and weight will be equal. Therefore we can use grams (g)
to measure both weight and mass on Earth. Will's apple will have a mass of 130g and will also weigh
130g. Will uses a balance in his classroom and finds that the mass of this apple is 130 g. If he finds the
weight of the same apple using a scale in his classroom, what will the weight of
the apple be?

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Baker – Science

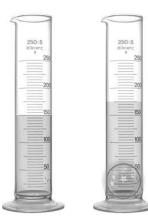
The apple will weigh _____ grams.

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If James takes the app mass weight	ole to the top of a mountair	n, which measuremer	nt will cha	ange?	
less gravity will be pulling pull on the apple.	ll change a very small amount at on it. An apple will weigh less be ntain, the apple will weigh (cause the force of gravity	y will have	a weaker ´	
	g, Patch, on a diet. That diet will ne of the material that makes up osing mass."				
by taking him soby taking him so	Patch to lose weight, how mewhere with less gravity mewhere with more gravit mewhere with the same ar	у	happen	?	
talks about losing or gainin the same place, mass is re	all Robin must do is take him son ng weight, he or she usually mea lated to weight. If Patch's mass null on something with a smaller	ns losing or gaining mass decreases, so will his wei	s. When m	easured in	
If Patch's diet goes well and he loses mass, what will happen to his weight? It will increase. It will also decrease. It will stay the same.					
DAY #2 - DIRECTIONS	: Read each passage and co	mplete the activities	after eac	h.	
centimeters, which are also its length by its width by it space that the object fills u cannot easily measure, you way with water. First, you graduated cylinder. Then, y record the new level of the	of the amount of space taken up called milliliters. To find the volume of an object of the colume of an object can place it in a graduated cylimust know how much water is a you place the object in the water water. Finally, you subtract the punt. This will give you the volume.	lume of a rectangular obj I amount of ct that you nder filled part Iready in the c. Next, you original amount			
What is the measure o	of how much space an object	ct takes up?			
	the block's length, width, and h lume of larger objects can be me			ill help find	
Select the measureme know in order to find t	ents that you need to the volume of this crate:			↑	
X X	= m³	5 m	11	2 m	
		3	K		

How can we find the volume of this rubber ball?

- by putting it in a graduated cylinder filled with a known amount of water
- by putting it in a graduated cylinder that does not have water
- by subtracting its length, width, and height
- by multiplying its length, width, and height

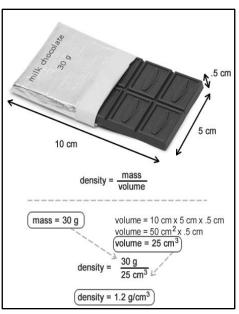




The length, width, and height of this ball would be very difficult to measure correctly. Therefore, we should place the ball in a graduated cylinder with a known amount of water in it. This will help us figure out how much space the ball takes up. The original amount of water in the graduated cylinder was 150 mL. After we put the ball in, the water rose to 175 mL. Subtracting the original amount of water from the final amount gives us the volume of the rubber ball. The ball has a volume of 25 mL.

Density (D) is the amount of mass a specific volume of an object has. To find the density, you must know the mass and volume of an object. Then, you divide the mass by the volume to get the density. Density is usually measured in grams per cubic centimeter (g/cm3) or grams per milliliter (g/mL).

The density of an object does not change depending on how much of it you measure. This is because the mass and volume of an object do change if the amount of the object you are measuring changes. However, when you divide the mass by the volume, it will always equal the same amount. Density equals the amount of matter within an object (mass) divided by the space it takes up (volume).





This stick of butter has a mass of grams.

This stick of butter has a volume of cm3.

If we want to find the density of this stick of butter, which should we equation

 $D = 113 g \div 54 cm^3$ $D = 54 \text{ cm}^3 \div 113 \text{ g}$ use?

 $D = 113 \text{ g x } 54 \text{ cm}^3$

 $D = 54 \text{ cm}^3 \text{ x } 113 \text{ g}$

To find the density of an object, divide its mass by its volume. The density of this butter is about 2.09 g/cm3. This rock has a mass of 240 grams and a volume of 80 cm3.

What is the density of the rock?

To find the density of an object, divide its mass by its volume.



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	The density of the tomato will not change if it is cut in half. The mass will change, and the volume will change, but dividing the new measurements will still equal the same density measurement. For example, if the volume of the whole tomato is 20 cm3, and the mass of the whole tomato is 200g, the density will equal 10 g/cm3. Cut in half, the volume of the tomato will be 10 cm3, the mass is 100g, and the density will still equal 10 g/cm3.						
□ It □ It □ It □ It	ill happen if this is mass will stay is will decorate will stay in mass will decorate density will stay is density will decorate will decorate will decorate will decorate will decorate.	crease by half. y the same. ease by half. y the same.					
DAY #3	- DIRECTIONS: F	Read each passage and complete the activiti	es after ea	ch.			
more bud liquid. Ma is denser	Buoyancy describes whether an object sinks or floats in another substance. Matter that floats in a liquid is more buoyant than the liquid. This is because the matter has a lower density, or is less dense, than the liquid. Matter that sinks in a liquid is less buoyant than the liquid. Sinking matter has a higher density, or is denser, than the liquid. Whether an object sinks or floats in another material is called its buoyancy. The density of an object compared to the density of the material it is in determines the object's buoyancy.						
more bu	uoyant, it is	_ tells us whether or not an object will float	t. When an	object is			
D = .87 g/mL	D = .78 g/mL	An object will float if its density is less than the density of the material it is in. The helium must be less dense than the air it is in. This causes the balloon to float. The density of all pure water is 1 g/mL.					
3 Joyne		Which of these objects would sink in a tub of water?					
Wendell has figured out the volume and mass of this log. Use his measurements to calculate the density of the log.							
		The density of the log is g/mL		00 grams : 1000 mL			
Will the	log be buoyant i	n pure water? YES / NO					
The log's density is 0.7 g/mL, which is less than 1 g/mL. Therefore, it will be buoyant, which means it will float in water.							
Some materials are very good for moving, or conducting, energy. Conductivity measures the ability of a material to move energy. Most metals are good conductors. That is why wires made out of metals, like copper, are used to help electricity flow from an outlet to a light bulb. Most metals are also good at carrying heat energy. That is why many pots and pans are made of metal. Water can also be a conductor. That is why it is dangerous to go swimming in a thunderstorm.							
Some materials have very low conductivity, like rubber and silicone. These materials are used in objects that insulate conductors, so they are called insulators. For example, metal wires are wrapped in rubber tubes to keep electricity inside the wire so that people do not shock themselves or start a fire from a spark. People often use silicone for cooking utensils so that the utensils do not melt when touching a hot							

pan.

Conduct	ty moscuroc	the ability of a ma	torial to
۱۲	ty measures	s the ability of a ma	teri

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Materials with high conductivity mo This means that they move energy	ove energy easily. Metal wires and water quickly and easily.	- both have high co	onductivity.
Select the materials with high	h conductivity:		
	cal charge to the doorknob, so her body over 60% water, and water is a good cor		
When Lane rubs her feet against the bedroom door, her touch creates a	ne carpet, static electricity builds up. If s small shock.	he touches the do	orknob on her
Which of the following must I Humans cannot be con Humans can be conduct Lane is made of metal.	ductors. ctors.		
The rubber around Jessica's of unsafe for Anita to continue of	computer cord is split open. Expla using the broken cord.	in one reason v	why it is
The rubber insulates the cord, which could shock Jessica, or it might spa	ch carries electricity. Without the insulati ark and start a fire.	ion, the electricity	in the wire
measure of how much of one substance can dissolve example, a teaspoon of sugar will of	ompletely, when stirred into another subtance can dissolve in another. Higher sole in another. Higher solute in another. However, even highly soluted in a cup of tea, but if you keep puld not be able to dissolve any more sugue bottom of the teacup.	lubility means that ble materials have bouring sugar into	t a larger a limit. For the tea,
	substance will dissolve in another. Solu se only certain matter dissolves in certa t dissolve in alcohol.		
- 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1			

Solubility is the measure of how much one substance can:

□ Dissolve in another

☐ Melt in another

□ Dissolve in itself

A good example of solubility comes from our oceans. The salt is soluble in the water because it is dissolved. The sand does not dissolve in the ocean water, so it is not soluble.

Ellie's aunt uses powdered laundry detergent. The powder is a substance that dissolves in water. One day, Ellie decides to help her aunt with the laundry. However, when she pulls her clothes out of the washer, they have some white powder on them. It smells fresh and clean, and Ellie realizes that the powder is laundry detergent.

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So S	What has happened? Some of the detergent was not soluble due to the amount of water in the washer. Ellie did not put enough laundry detergent in the washer. All of the detergent was soluble due to the amount of water in the washer. Ellie put too much laundry detergent in the washer. Ellie must have put too much detergent in the washer. The solubility of that amount of detergent was too low compared to the amount of water in the washer. Therefore, not all of the detergent in Ellie's laundry dissolved in the water.					
	oda and honey both have hig y are stirred into water.	gh solubility in water. Both substances	s dissolve, or mix co	ompletely,		
		soluble in water that were not r substances are soluble.	named in this le	sson.		
1						
DAY #4	DIRECTIONS: Review D	ays 1-3 by marking all the corre	ct answers.			
What is	a physical property?					
□ th	e measure of the amou	ınt of matter in an object				
		ınt of matter in a given volume	of an object			
	e measure of the pull of					
	_	rial that can be measured or det	tected by the se	nses		
_	is					
		int of matter in an object				
	hether an object sinks					
	e measure of the pull of		_			
		ınt of space taken up by an obje	ect			
What is						
	e ability of a material t					
	hether an object sinks					
		int of matter in an object				
		ınt of space taken up by an obje	ect			
	volume?	t of our object				
	e measure of the heigh					
		int of energy in a material	. in anathau			
		h of one substance can dissolve				
		ınt of space taken up by an obje	3CL			
	is	and length of an object				
		h of one substance can dissolve	in another			
		int of energy in a material	in another			
		int of matter in a given volume	of an object			
	buoyancy?	ant of matter in a given volume	or arr object			
	e ability of a material t	o carry energy				
	hether an object shrink					
	hether an object sinks	_				
	-	h of one substance can dissolve	in another			
	civity is					

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		, 5,	other	
What	is solubility?			
	the measure of how much	ch of one substance can dissolve in an	other	
	the ability of a substance	e to melt completely when heated		
	the measure of how much	ch energy it takes to melt a substance	!	

DAY #5 - DIRECTIONS: Review and complete the activities after each.

□ the ability of a substance to turn into water vapor



Use your five senses to describe at least three physical properties of this marshmallow.

This car has a mass of 1360 kg and weighs 1360 kg. If taken onto the Moon, where the force of gravity is less, what will happen to the car?

- □ Its mass will increase.
- □ Its mass will decrease.
- ☐ Its weight will increase.
- ☐ Its weight will decrease.





What is the volume of this box? (hint: $L \times W \times H = V$)

CIRCLE the things that you would need in order to calculate the volume of a golf ball.



Bake	r – Science	Student Name:	Per:	Page 8
rock densi	collection. T ty? the empty	shoeboxes that are exactly the same the other shoebox is empty. Which of shoebox is full of rocks		
		ems would be buoyant in this water? ace above or below the water.	Which would not? Mo	ve the items
		D= 7.2 g/mL	D= .42 g/mL	
Label	the insulate	or and label the conductor in this picto	ure:	
•		ductor because it moves heat from the ecause it keeps heat from moving to t		he potholder
powd	er into warr ollowing stat	not chocolate from her favorite hot che m milk. Most of the powder dissolves, tements are true?	but some of it does i	not. Which of
	The powder the powder	r is soluble in the milk, but there was ·.	not enough milk to c	lissolve all of
	of the milk			
	of the milk	soluble in the powder, but there was . r was not soluble in the milk.	not enough powder t	o dissolve all
	Which	of the following are physical properties smooth wooden brown all the above	es of this rocking cha	ir?
	nass? Check Mass chang Weight onl Weight and Weight cha	owing statements accurately describe all that are true. ges at the top of a mountain. y changes when mass changes. I mass are the same thing. Inges depending on gravity. The same everywhere in the universe		een weight

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measure the letmeasure the wmeasure the hadd the measure	vidth		
volume of this comb Fill a graduate Record the lev Find the difference graduated cyline	ed to do in order to find the? Check all that are true. d cylinder with a known amount of wa el of water after the comb is placed in ence between the amount of water aft nder and the original amount of water livide the length, width, and height of	ater. I the graduated cyli Per the comb is add I	
	Quinton knows that the volume of his mL. What does he need to do to find Check all that are true. He needs to use a balance to find the He needs to use a scale to find the He needs to divide the mass by the He needs to divide the weight by the	the density of the the mass of the con e weight of the cont e volume of the con	controller? ntroller. troller. ntroller.
Which of the followir Check all that are true It will float bed The marble will Ihe The marble will	e with a density of 2.76 g/mL. The dering are true about the marble? ue. cause it is more dense than water. Il be buoyant in the water. Il not be buoyant in the water. ause it is more dense than water.	nsity of pure water	is 1 g/mL.
wires poking out of to coming to clear the place if the metal wire the black rubb the black rubb	ped in half during a storm in Kylie's not the black rubber tube that covers the power lines away. Which part of the post when moving the power line? In spoking out, because they insulate eler tube, because it conducts electricity or tube, because it insulates electricity is poking out, because they conduct electricity.	power lines. Firefig ower line should th lectricity y y	hters are
hot water. The salts	baths. She has bath salts and bath of dissolve completely in the water, but g with the water. Which substance is solve	the oil swirls aroun	